

Upconversion Properties of Er³⁺/Yb³⁺ Co-doped TeO₂-TiO₂-K₂O Glasses

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The Er³⁺/Yb³⁺ co-doped TeO₂-TiO₂-K₂O glasses were prepared by conventional melting procedures, and their upconversion spectra were performed. The dependence of luminescence intensity on the ratio of Yb³⁺/Er³⁺ was studied, and the relationship between green upconversion luminescence intensity and Er³⁺ concentration is discussed in detail. The 546 nm green upconversion luminescence intensity is optimised in the studied glasses either when the Yb³⁺/Er³⁺ ratio is 25/1 and Er³⁺ concentration is 0.1 mol%, or when the Yb³⁺/Er³⁺ ratio is 10/1 and Er³⁺ concentration is 0.15 mol%. These glasses could be one of the potential candidates for LD pumping microchip solid-state lasers.

KEY WORDS : Upconversion; Rare earth ions; Tellurite glasses.

INTRODUCTION

Upconversion properties of rare earth ions-doped glasses have attracted much attention in recent years due to their excellent optical properties. These materials have potential applications in the information process, data storage, display technology, undersea transmission, and medicine [1–6]. Particularly, tellurite glasses have been investigated extensively because of their relative low phonon energy, high refractive index, good corrosion resistance, and high thermal stability.

Er³⁺/Yb³⁺ co-doped glasses [7,8] are attractive because Yb³⁺ has excellent absorption efficiency [9] at 976 nm, and Er³⁺ possesses an intermediate level (⁴I_{11/2}) with long life-time. This work is to study the influence of Er³⁺ concentration and the Er³⁺/Yb³⁺ ratio in tellurite glasses on the 546 nm green upconversion luminescence intensity, and to find out better candidate for the microchip short wavelength lasers.

The compositions of the Er³⁺/Yb³⁺ co-doped tellurite glasses are shown in Table I. Their absorption and upconversion spectra were measured. Two green emission bands centred at 523 and 547 nm, corresponding to the ²H_{11/2}→⁴I_{15/2} and ⁴S_{3/2}→⁴I_{15/2} transitions, respectively, and one red emission band centred at 657 nm by ⁴F_{9/2}→⁴I_{15/2} transition, were observed. The dependence of luminescence intensity on the ratio of Yb³⁺/Er³⁺ and Er³⁺ concentration are discussed in detail.

EXPERIMENTS

Glasses Preparation

All starting materials for Er³⁺/Yb³⁺ co-doped glasses are analytical grade chemicals of TeO₂, TiO₂, K₂CO₃, Er₂O₃, and Yb₂O₃. The glasses base compositions (in mol%) are 80TeO₂-(10-χ-χR)TiO₂-10K₂O-χEr₂O₃-χRYb₂O₃ (χ = 0.1, 0.15, 0.2, 0.25, 0.3, 0.4, 0.5, and 1.0; R, the ratio of Yb³⁺/Er³⁺, R = 1/10, 1/5, 1/4, 1/3, 1/2, 1/1, 2/1, 3/1, 4/1, 5/1, 10/1, 15/1, 20/1, 25/1, 30/1, and 35/1), and the studied compositions are listed in Table I.

The well-mixed batches were melt in Pt crucible at 850°C for 30 min in an electric furnace, then the glasses were poured into a stainless steel mould and annealed in a

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Table I. The Glasses Compositions (in Molar Percentage)

Sample no	TeO ₂	K ₂ O	TiO ₂	Er ₂ O ₃	Yb ₂ O ₃	χ	R
Series I							
TTK-1-1	80	10	8.9	1.0	0.1	1.0	1/10
TTK-1-2	80	10	9.4	0.5	0.1	0.5	1/5
TTK-1-3	80	10	9.5	0.4	0.1	0.4	1/4
TTK-1-4	80	10	9.6	0.3	0.1	0.3	1/3
TTK-1-5	80	10	9.7	0.2	0.1	0.2	1/2
TTK-1-6	80	10	9.8	0.1	0.1	0.1	1/1
TTK-1-7	80	10	9.7	0.1	0.2	0.1	2/1
TTK-1-8	80	10	9.6	0.1	0.3	0.1	3/1
TTK-1-9	80	10	9.5	0.1	0.4	0.1	4/1
TTK-1-10	80	10	9.4	0.1	0.5	0.1	5/1
TTK-1-11	80	10	8.9	0.1	1.0	0.1	10/1
Series II							
TTK-2-1	80	10	8.9	0.1	1.0	0.1	10/1
TTK-2-2	80	10	8.4	0.1	1.5	0.1	15/1
TTK-2-3	80	10	7.9	0.1	2.0	0.1	20/1
TTK-2-4	80	10	7.4	0.1	2.5	0.1	25/1
TTK-2-5	80	10	6.9	0.1	3.0	0.1	30/1
TTK-2-6	80	10	6.4	0.1	3.5	0.1	35/1
TTK-2-7	80	10	8.35	0.15	1.5	0.15	10/1
TTK-2-8	80	10	7.8	0.2	2.0	0.2	10/1
TTK-2-9	80	10	7.25	0.25	2.5	0.25	10/1
TTK-2-10	80	10	6.7	0.3	3.0	0.3	10/1

muffle oven at their T_g points. The annealed glasses were cut into 15 mm × 15 mm × 1.5 mm pieces and then polished for optical measurements.

Properties Measurements

The absorption spectra of the glass samples were measured with an American Perkin-Elmer-Lamda 900 UV/VIS/NIR spectrophotometer. Upconversion luminescence measurements were performed by a

976 nm LD pumping and a Triax-320 spectrophotometer detecting.

All the measurements were done at room temperature.

RESULTS AND DISCUSSION

Absorption Spectra

In the absorption spectra measured for all the samples shown in Fig. 1, it can be seen clearly that there are seven absorption bands, their absorption peaks centered at 486, 521, 546, 657, 808, 977, and 1533 nm, respectively. The energy levels diagram of Er³⁺/Yb³⁺ is shown in Fig. 2, referring to W. T. Carnall's work [10], the data of the energy levels' wave-numbers were acquired from the absorption spectra. The seven absorption bands are corresponded to the transitions of ⁴I_{15/2}(Er³⁺) → ⁴F_{7/2}(Er³⁺), ⁴I_{15/2}(Er³⁺) → ²H_{11/2}(Er³⁺), ⁴I_{15/2}(Er³⁺) → ⁴S_{3/2}(Er³⁺), ⁴I_{15/2}(Er³⁺) → ⁴F_{9/2}(Er³⁺), ⁴I_{15/2}(Er³⁺) → ⁴I_{9/2}(Er³⁺), ⁴I_{15/2}(Er³⁺) → ⁴I_{11/2}(Er³⁺) and ²F_{7/2}(Yb³⁺) → ²F_{5/2}(Yb³⁺), and ⁴I_{15/2}(Er³⁺) → ⁴I_{13/2}(Er³⁺). There exists the absorption peak at 977 nm, which is resonance absorption with 976 nm incident light. Therefore, we used 976 nm LD to pump the samples. Compared with the 977 nm absorption bands of Er³⁺ single doped material showing in Fig. 3, the corresponding absorption bands of Er³⁺/Yb³⁺ co-doped matrix were more intense, this is because Er³⁺/Yb³⁺ co-doped matrix has a larger absorption cross-section areas. As shown in Fig. 1a, the 977 nm absorption bands became more and more predominant in the sequence of TTK-1-*n* (*n* = 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, and 11). That is to say, when the ratio of Yb³⁺/Er³⁺ increases from 1/10 to 10/1, the

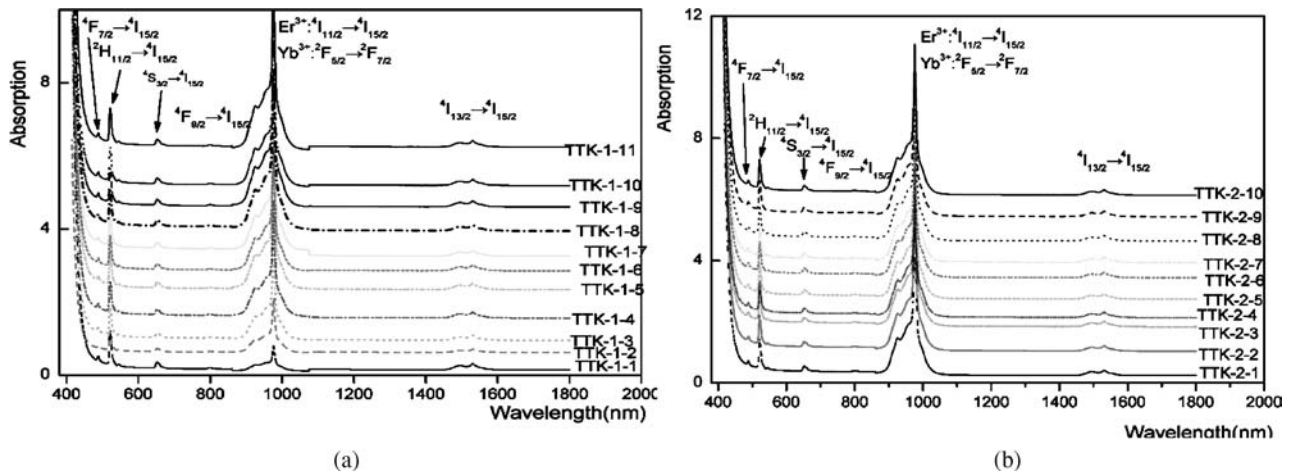


Fig. 1. The absorption spectra of the studied glasses. (a) TTK-1-*n* (*n* = 1, 2, 3, 4, 5, 6, 7, 8, 9, 10, and 11). (b) TTK-2-*n* (*n* = 1, 2, 3, 4, 5, 6, 7, 8, 9, and 10).

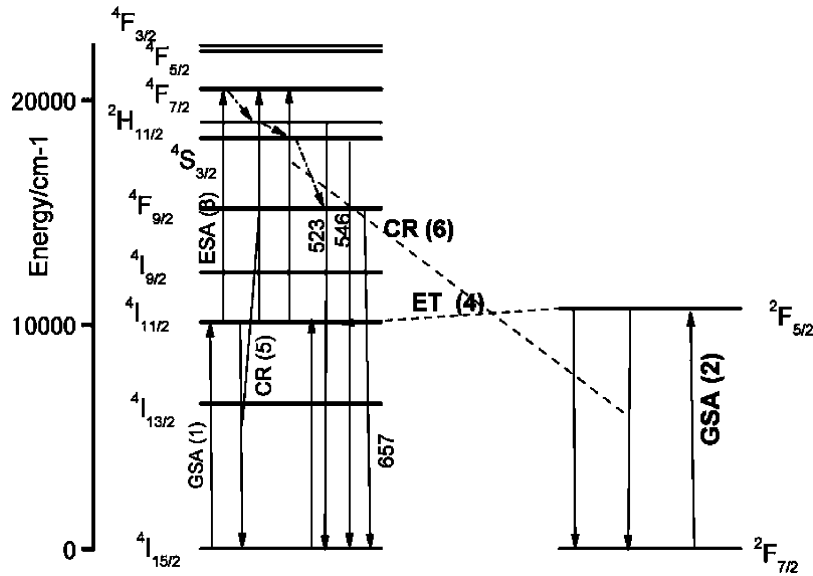


Fig. 2. The simple energy level diagram of Er³⁺/Yb³⁺ co-doped TTK glasses.

977 nm absorption band gets more and more significant as the Er³⁺ concentration remains fixed; this is due to the transition of ²F_{7/2}(Yb³⁺)→²F_{5/2}(Yb³⁺) with high absorption efficiency. Taken TTK-1-10 and TTK-1-11 for an example, the absorption integral area (46.57 mm²) and absorption peak height (1.414 mm) of TTK-1-11 are larger than those of TTK-1-10 (44.87 mm², 1.407 mm).

However, as for the absorption spectra of the TTK-2-*n* (*n* = 1, 2, 3, 4, 5, 6, 7, 8, 9, and 10) shown in Fig. 1b, no change can be detected in the 977 nm absorption bands. This is due to the saturation absorption of Yb³⁺, as increasing the concentration of Yb³⁺ to some extent, the increasing absorption will be reach the saturation point, 1 mol% in our system.

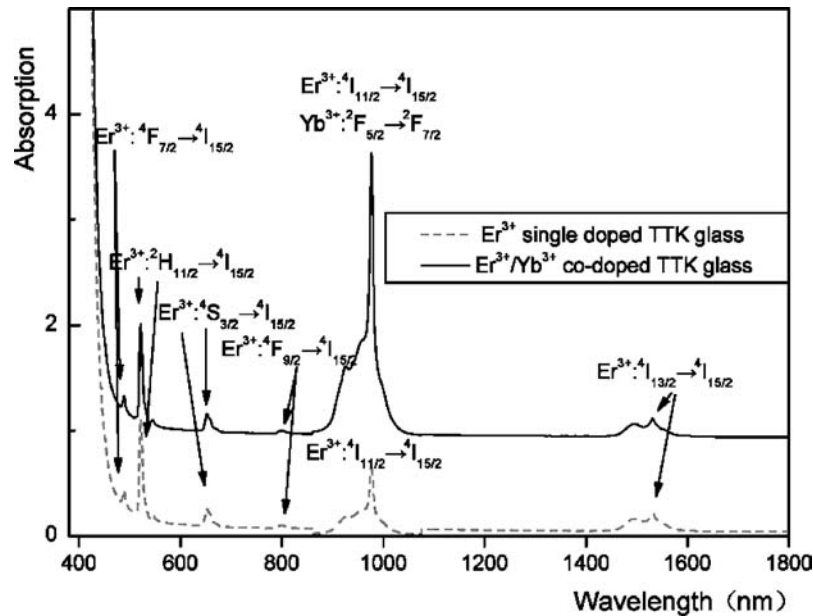


Fig. 3. The absorption spectra of Er³⁺/Yb³⁺ co-doped TTK glasses is compared with that of Er³⁺ single doped TTK glasses.

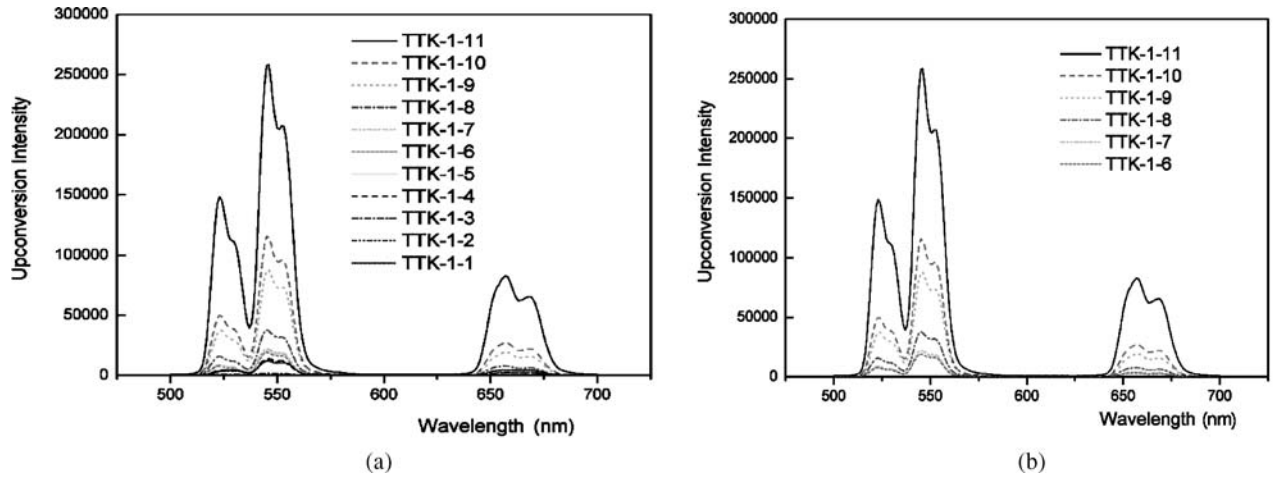


Fig. 4. Upconversion luminescence spectra of $\text{Er}^{3+}/\text{Yb}^{3+}$ co-doped TTK-1- n glasses excited by 800 mW LD. (a) TTK-1- n ($n = 1, 2, 3, 4, 5, 6, 7, 8, 9, 10,$ and 11). (b) TTK-1- n ($n = 6, 7, 8, 9, 10,$ and 11).

Upconversion Spectra

The upconversion spectra of series I (TTK-1- n) are shown in Fig. 4. There are three emission bands in the visible light range, 523, 546, and 657 nm, respectively.

Figure 5 shows the intensity of 546 nm emission luminescence as a function of the excitation power, the plots are in $\log(I_{\text{up}})$ versus $\log(I_{\text{IR}})$. These results suggested that the $^2\text{H}_{11/2}$ level was populated by a two-photon steps process. The upconversion processes [8,11,12] can be de-

scribed as below:

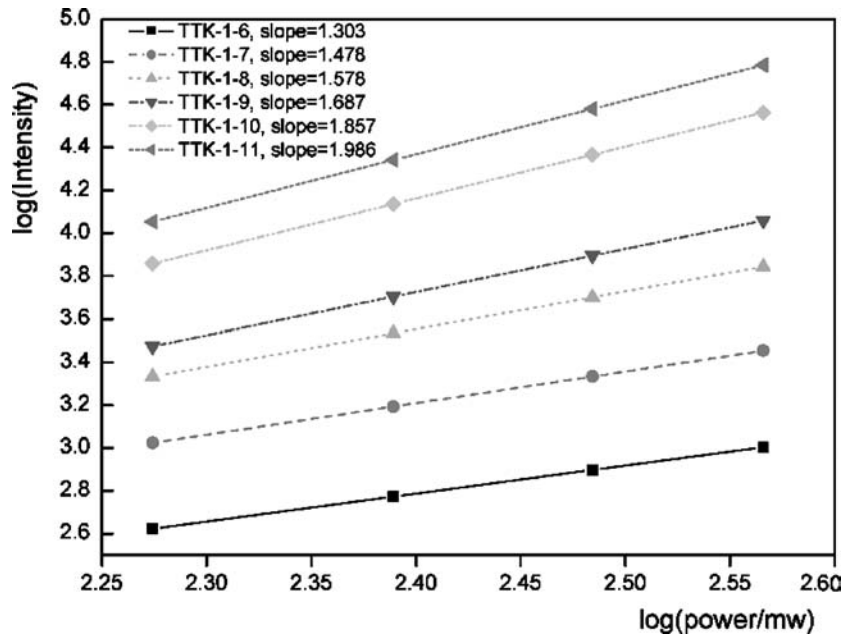
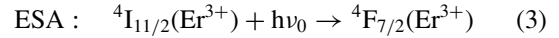
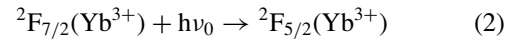
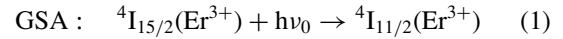


Fig. 5. The 546 nm upconversion luminescence intensity of TTK-1- n ($n = 6, 7, 8, 9, 10,$ and 11) VS pump power. The plot is in $\log(I_{\text{up}})$ vs. $\log(I_{\text{IR}})$.

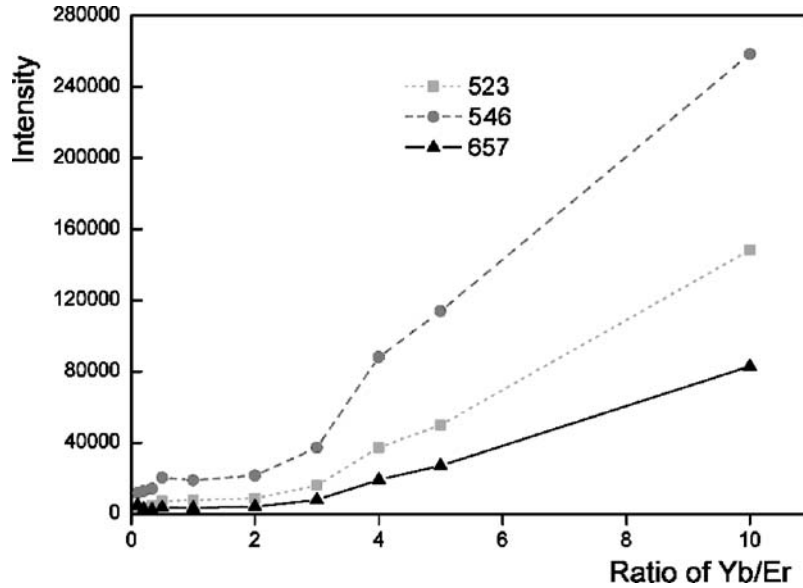
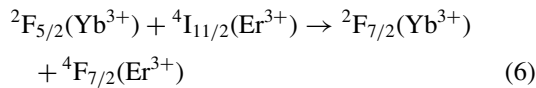
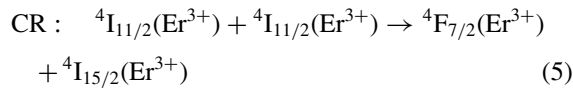
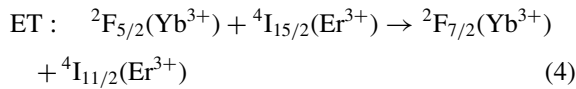


Fig. 6. The dependence of upconversion luminescence intensity on R (presentation for ratio of Yb³⁺/Er³⁺).



As discussed in Section Absorption Spectra, the absorption at 977 nm of Yb³⁺ gets more and more predominant with increasing ratio of Yb³⁺/Er³⁺. This means that the ET (energy transfer) mechanism will be more dominant, that is to say, the process [Eqs. (2), (4), and (6)] will be more dominant when the ratio of Yb³⁺/Er³⁺ increases. Referred to Fig. 2 (the above-mentioned process are all marked in Fig. 2), due to the resonate absorption at 976 nm, Yb³⁺ strongly absorbed the incident photons of 976 nm LD, a Yb³⁺ ion rose from ground state ²F_{7/2} to the excited state ²F_{5/2}, then the excited Yb³⁺ transferred its' energy to Er³⁺ by ET (4) process as Eq. (4), thus caused the Er³⁺ ion to excite from its' ground state ⁴I_{15/2}

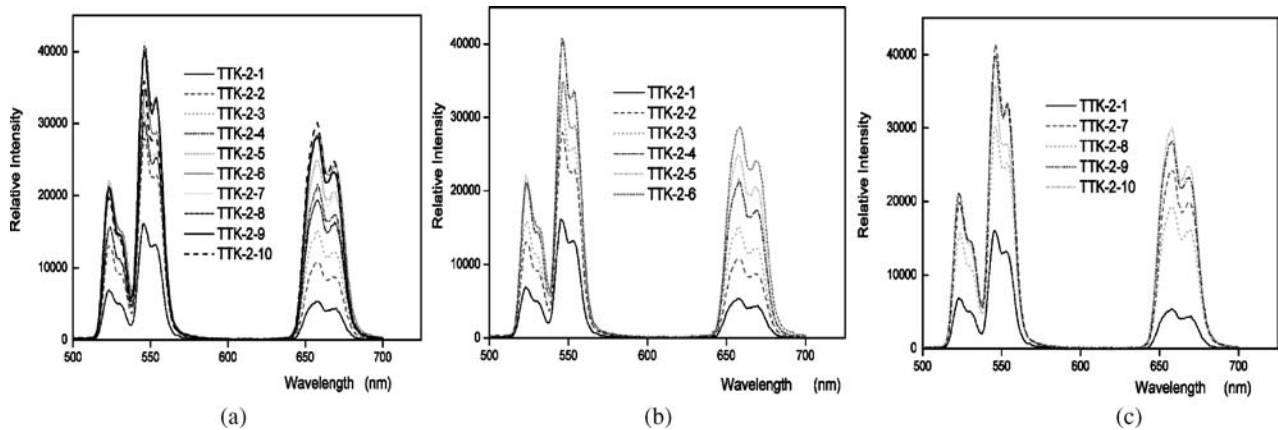


Fig. 7. Upconversion luminescence spectra of Er³⁺/Yb³⁺ co-doped TTK-2- n glasses excited by 490 mW LD. (a) TTK-2- n ($n = 1, 2, 3, 4, 5, 6, 7, 8, 9$, and 10); (b) TTK-2- n ($n = 1, 2, 3, 4, 5$, and 6); and (c) TTK-2- n ($n = 1, 7, 8, 9$, and 10).

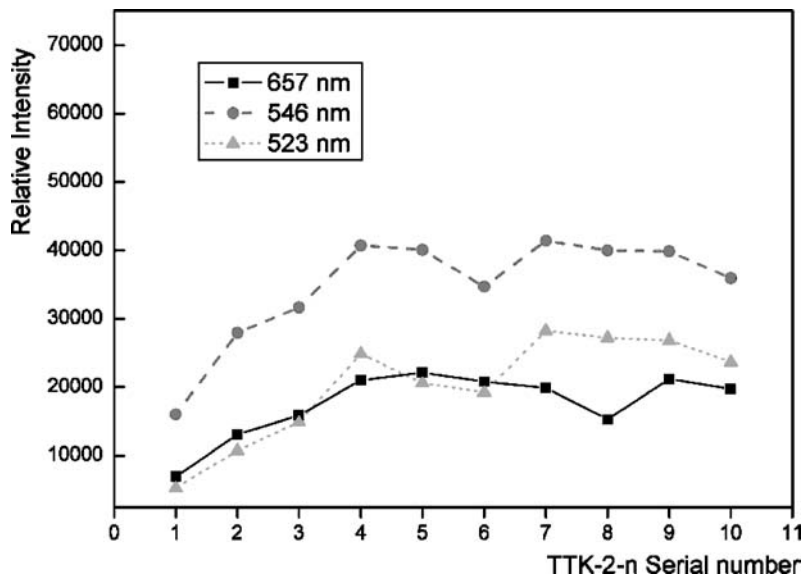


Fig. 8. Dependence of upconversion luminescence intensity on $\text{Yb}^{3+}/\text{Er}^{3+}$ ratio and Er^{3+} concentration. There are two maximum, TTK-2-4 and TTK-2-7.

to the excited state $^4\text{I}_{11/2}$, the Er^{3+} ion in $^4\text{I}_{11/2}$ could be further excited to $^4\text{F}_{7/2}$ through the CR (6) process. Since the life-time of the Er^{3+} ions at $^4\text{F}_{7/2}$ level is too short, it relaxes rapidly to the $^2\text{H}_{11/2}$ via non-radioactive transition. Similarly, those Er^{3+} ions at $^2\text{H}_{11/2}$ level may also non-radioactively relax to $^4\text{S}_{3/2}$ level, and then the Er^{3+} ions at $^4\text{S}_{3/2}$ level would non-radioactively relax to $^4\text{F}_{9/2}$ level. When the ions in $^2\text{H}_{11/2}$ level radiate relax to the ground level $^4\text{I}_{15/2}$, they emit the 523 nm green luminescence; similarly, the ions in level $^4\text{S}_{3/2}, ^4\text{F}_{9/2}$ relax to the ground level $^4\text{I}_{15/2}$, they respectively emit 546 and 657 nm upconversion luminescence.

The relationship between the upconversion luminescence intensity and the ratio of $\text{Yb}^{3+}/\text{Er}^{3+}$ are shown in Fig. 6; the 523, 546, and 657 nm emission intensities in-

crease as R increases. Especially, from TTK-1-6 to TTK-1-11, the upconversion luminescence intensity increase rapidly with increasing R . From Figs. 4b and 6, the increasing trend is obvious.

In order to get the optimal Er^{3+} concentration and $\text{Yb}^{3+}/\text{Er}^{3+}$ ratio for the potential application in microchip solid-state lasers, Series II experiments were carried out. The upconversion spectra of TTK-2- n are shown in Fig. 7. Figure 7b shows the dependence of upconversion intensities on R , where the concentrations of Er^{3+} were fixed at 0.1 mol% for all samples from TTK-2-1 to TTK-2-6. The 546 nm upconversion peak intensities were in the order: TTK-2-1 < TTK-2-2 < TTK-2-3 < TTK-2-6 < TTK-2-5 < TTK-2-4. From the experiment the maximum intensity was obtained around $R = 25/1$ in our studied glass system.

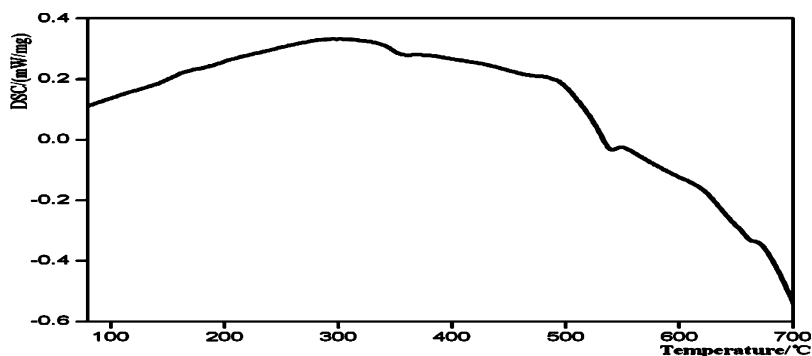


Fig. 9. The DSC curve of TTK-2-7.

The explanation of the above phenomenon can be given as: The Yb³⁺ concentration will increase as R increases, and the so-called concentration quenching [13] of Yb³⁺ would happen; therefore, the energy transfer will be restricted by some concentration limit, which induces the upconversion luminescence decrease.

The relationship between upconversion intensity and Er³⁺ concentration are clearly shown in Fig. 7c, where all samples (TTK-2-1, TTK-2-7, TTK-2-8, TTK-2-9, and TTK-2-10) were set to $R = 10/1$, it revealed the upconversion intensity order: TTK-2-1 < TTK-2-10 < TTK-2-9 < TTK-2-8 < TTK-2-7. The phenomenon can be interpreted as follows: As concentration of Er³⁺ increases, the Er³⁺ ions play the main role in determining upconversion luminescence intensity. While the Er³⁺ concentration increases to some extent, Er³⁺-Er³⁺ interaction will happen. This interaction will transfer their energy in thermal form, thus lower luminescence intensity. This is so-called concentration quenching.

The maximum of the upconversion intensity was obtained from two ways shown in Fig. 8; one is fixing Er³⁺ concentration at 0.1 mol% and changing the Yb³⁺/Er³⁺ ratio, TTK-2-4; the other is fixing the Yb³⁺/Er³⁺ ratio at $R = 10/1$ and changing Er³⁺ concentration, TTK-2-7. The two maximum intensities are very close in quantity in our work.

In order to meet the application need of the studied material, the thermal quality of TTK-2-7 was also studied, according to Fig. 9, there is no crystalline peak in the whole thermal process. That is to say, the thermal quality of TTK-2-7 is excellent.

CONCLUSIONS

The absorption spectra and upconversion spectra of Er³⁺/Yb³⁺ co-doped TTK glasses are measured and discussed. The dependence of upconversion luminescence intensity on the Er³⁺ concentration and the ratio of Yb³⁺/Er³⁺ are studied in detail. The optimal upconversion intensity can be obtained either when the Yb³⁺/Er³⁺ ratio is 25/1 and Er³⁺ concentration is 0.1 mol%, or when the Yb³⁺/Er³⁺ ratio is 10/1 and Er³⁺ concentration is 0.15 mol%. The intense green upconversion luminescences were observed in all the samples with naked eyes, which announce their potential application advantage in the field of LD pumping microchip solid-state lasers.

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